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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/562,435	12/28/2005	Mitsunori Nodono	023174-0151	3862
22428 7590 03/03/2010 FOLEY AND LARDNER LLP			EXAMINER	
SUITE 500		SIDDIQUEE, MUHAMMAD S		
3000 K STREE WASHINGTO			ART UNIT	PAPER NUMBER
			1795	
			MAIL DATE	DELIVERY MODE
			03/03/2010	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/562,435	NODONO, MITSUNORI			
Office Action Summary	Examiner	Art Unit			
	MUHAMMAD SIDDIQUEE	1795			
The MAILING DATE of this communication app Period for Reply	pears on the cover sheet with the c	orrespondence address			
A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D - Extensions of time may be available under the provisions of 37 CFR 1.1 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION (36(a). In no event, however, may a reply be time will apply and will expire SIX (6) MONTHS from the cause the application to become ABANDONE	lely filed the mailing date of this communication. (35 U.S.C. § 133).			
Status					
1) ☐ Responsive to communication(s) filed on 15 D 2a) ☐ This action is FINAL . 2b) ☐ This 3) ☐ Since this application is in condition for alloware closed in accordance with the practice under B	s action is non-final. nce except for formal matters, pro				
Disposition of Claims					
4) ☐ Claim(s) 1-16 is/are pending in the application 4a) Of the above claim(s) 11-16 is/are withdray 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-10 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/o Application Papers 9) ☐ The specification is objected to by the Examine 10) ☐ The drawing(s) filed on 28 December 2005 is/a Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct	wn from consideration. or election requirement. er. are: a)⊠ accepted or b)□ object drawing(s) be held in abeyance. See	e 37 CFR 1.85(a).			
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.					
Priority under 35 U.S.C. § 119					
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 					
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 12/28/2005, 1/11/2008.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ite			

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filed on 10/15/2009.

DETAILED ACTION

Election/Restrictions

Claims 11-16 are withdrawn from further consideration pursuant to 37 CFR
 1.142(b), as being drawn to a nonelected invention, there being no allowable generic or linking claim. Applicant timely traversed the restriction (election) requirement in the reply

Applicant's election with traverse of Group I in the reply filed on 10/15/2009 is acknowledged. Applicant's arguments with respect to the traversal have not been found persuasive. Applicant argues that since the inventions are related to polymer electrolyte, the subject matter is not divergent and would not be a serious burden to the examiner.

Examiner respectfully disagrees. Since the inventions are classified in different classes and requires searches in different classes. Therefore, the search for the two separate groups is not coextensive and would thus place a burden on the Examiner.

The requirement is still deemed proper and is therefore made FINAL.

Claim Objections

- 2. Claim 1 is objected to because of the following informalities: Claim 1 recites "....fine pores which is *fill* with...". Appropriate correction is required.
- 3. Claim 10 is objected to because of the following informalities: Claim 10 recites to depend from claim 9; however, it should depend from claim 8. Appropriate correction is required.

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Claim Rejections - 35 USC § 103

4. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

- 5. The factual inquiries set forth in *Graham* **v.** *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.
 - 2. Ascertaining the differences between the prior art and the claims at issue.
 - 3. Resolving the level of ordinary skill in the pertinent art.
 - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 6. Claims 1-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Uchida et al (US 2002/0182478 A1).

Regarding claims 1-2, Uchida discloses a fuel cell comprising a polymer electrolyte membrane and a pair of electrodes having a catalyst layer on a surface which is in contact with the polymer electrolyte membrane and sandwiching the polymer electrolyte membrane therebetween, wherein the catalyst layer of at least one of the electrodes comprises carbon particles supporting a noble metal catalyst, and the carbon particles include at least two kinds of carbon particles adsorbing a polymer electrolyte in mutually different dispersed states [paragraph 0014]. Here, the electrolyte membrane, the electrolyte and the carbon particles form the polymer electrolyte composite membrane [paragraph 0061]. Uchida also teaches that polymer electrolyte comprises a

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hydrophilic group and a hydrophobic group, and combined size of the hydrophilic and hydrophobic domains ranges from 10 nanometer to 200 nanometer. Uchida further teaches that pores formed on the composite membrane ranges from 10 to 200 nanometer. Hence, due to the size of the hydrophilic and hydrophobic domains and the pore diameter, it is possible to bring the catalyst particles and the polymer electrolyte into contact with each other satisfactorily and increase the reaction area of the catalyst [paragraph 0075-0079]. Thought, Uchida does not explicitly express the relationship of hydrophilic/hydrophobic domains size and the pore diameter, it is within the technical reach of a skilled artisan to present the relationship of the size of the hydrophilic and hydrophobic domains and the pore diameter in a mathematical formula.

Regarding claims 3-6, Uchida teaches that the size of the hydrophilic and hydrophobic domains ranges from 10 nanometer to 200 nanometer [paragraph 0075].

Regarding claims 7-9, Uchida teaches that the hydrophilic repeating unit has a (–SO₃H) cation-exchange group [formula 3; paragraph 0072-0073].

7. Claim 10 is rejected under 35 U.S.C. 103(a) as being unpatentable over Uchida et al (US 2002/0182478 A1) as applied in claims 1-9 and further in view of Inagaki et al (US 2003/0104284 A1).

Regarding claims 10, Uchida remains silent about a positively charged ionexchange group. However, Inagaki teaches that hydrophilic polar group can be an
anionic group such as a sulfonic group or a cationic group such as an amine [paragraph
0019]. Therefore, it is within the technical reach of a skilled artisan to choose an anionic
group or a cationic group which gives the electrolyte an ionic charge to function.

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Any inquiry concerning this communication or earlier communications from the examiner should be directed to MUHAMMAD SIDDIQUEE whose telephone number is (571) 270-3719. The examiner can normally be reached on Monday-Thursday, 7:30 am to 4:00 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Muhammad Siddiquee/ Examiner, Art Unit 1795

/PATRICK RYAN/ Supervisory Patent Examiner, Art Unit 1795